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Solution: $p_{\text{abs}} = 140 \text{ kPa} = 20.32 \text{ psia}$
 $14.70 \text{ psi} = 101.3 \text{ kPa}$
 $p_{\text{gage}} = p_{\text{abs}} - p_{\text{atm}} = (20.32 \text{ psia}) - (14.70 \text{ psia}) = 5.62 \text{ psi}$
 $p_{\text{gage}} = 5.62 \text{ psig c.}$

Situation: Pressure values need to be converted. Find: Calculate the

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~~Solution~~ absolute pressure (psia)
corresponding to a pressure of 0.55
bar (gage). Properties: $p_{atm} =$
14.70 psi: Solution: $p_{gage} = 0.55$ bar
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