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A chemical

equation

represents a

chemical

reaction. $Mg + O$

$2 \rightarrow MgO$. Since

the number of

atoms of each

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element is not the same on both the sides, then the equation is unbalanced

because the mass is not the same on both sides of the equation.

Such a chemical equation is a skeletal chemical equation for a

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this lesson,
you'll learn how
to balance a

chemical
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Reactions and
Equations: In
our day to day
life, we see
several chemical
changes take
place. To study
these changes
scientists

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Chemistry

Classified it in
different types
of chemical
reactions. In
this section, we
will read
chemical
reactions and
equations notes
for class 10.

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~~Reactions and~~

~~Equations Notes~~

Read Book Chemistry ~~for class 10~~

~~with PDF~~

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Equations show

the products

that the

combination of

reactants yield,

but the equation

doesn't show how

the reactants

form those

products. What

we don't see are

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the underlying
chemical and
physical changes
that allow
molecules to
rearrange to
change
properties or
create new
substances.

~~Net Ionic~~

~~Equations | Unit~~

~~4: Chemical~~

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will be awarded.

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submitted to the
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later than 11:59
pm Sunday,
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Reactions and

Equations 1.

Directions: 1.

Substitute

symbols and

formulas for

words. 2.

Predict the

products. 3.

Balance the

equation. 4.

Give the type of

reaction.

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~~Chemical Reactions And~~

~~Reactions and~~

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The integrated rate equation for a first-order reaction is: $[A] = [A]_0 e^{-kt}$. Where, $[A]$ is the current concentration of

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the first-order
reactant. $[A]_0$
is the initial
concentration of
the first-order
reactant. t is
the time elapsed
since the
reaction began.
 k is the rate
constant of the
first-order
reaction.

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~~First Order And~~

~~Reactions~~

~~(Chemical~~

~~Kinetics) Free~~

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Equation:

Representation

of chemical

reaction using

symbols and

formulae of the

substances is

called Chemical

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Equations And

Example: $A + B \rightarrow$

$C + D$ In this

equation, A and

B are called

reactants and C

and D are called

the products.

The arrow shows

the direction of

the chemical

reaction.

Condition, if

any, is written

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generally above
the arrow.

Reactions And
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...

Balanced
chemical
equation → The
chemical
equation that
shows the

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Reactions And

reaction needs
to be balanced.

A balanced

chemical

equation occurs
when the number

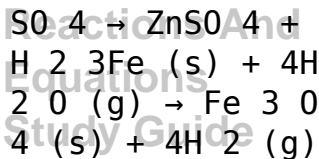
of the atoms

involved in the
reactants side

is equal to the
number of atoms

in the products
side. $Zn + H_2$

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~~Chemical~~

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A chemical

change is the

change in the

chemical

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composition of a substance like the formation of a precipitate in a chemical reaction. A physical change can be reversed to result in the same ...

~~Explain the difference between a~~

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Chemical
Formulas,

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Instructions.

Choose your
answers to the
questions and
click 'Next' to
see the next set
of questions.

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A chemical reaction involves the transformation of one or more substances into another substance or

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Substances. A chemical equation involves a change in composition and is represented by a chemical equation. A chemical equation provides a concise picture of a chemical change.

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~~Equation ...~~

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Chemical
equations are
different from
mathematical
equations
because the two

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Key

'before' and

'after' of a

chemical

reaction. In

mathematical

equations, an...

~~What is a~~

~~Chemical~~

~~Equation?~~

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Balancing a
chemical

equation before

performing the

chemical

reaction is

important

because it shows

the quantitative

relations of the

given reactants

and products

which are formed

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~~Why is it
important to
balance chemical
equations before~~

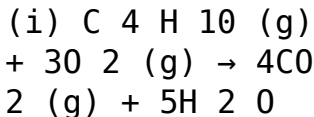
~~...~~

This is an
exothermic
reaction and
energy released
is used to cook.
Carbon di oxide
gas and steam is

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product for this reaction. Which of the following represents proper chemical equation for this process?



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Equations

In the decomposition reaction, 1 mole of water (MW = 18.015 g/mol) was produced for every mole of CuO (MW = 79.545 g/mol) produced. Given that 3.327 g of CuO was produced during

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the reaction, And

how many grams
of water were
released as

water vapor?

(number of moles
= mass (g) /
molar mass

(g/mol)). Choose
the closest

answer. Select

one: a. 0.92 g

b. 1.80 g

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~~Study Quiz 6:~~

~~Chemical~~

~~Reaction Types~~

~~and Their~~

~~Equations ...~~

The chemical equation will tell us the number of moles that are used, which in turn can tell us the mass and volume.
Moles in

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Chemical Reactions And
Equations Now
let's think
about the dozen
atoms again
in...

~~Using Moles in
Chemical
Equations |
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Another simple
way to express a
chemical

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Reaction is with a skeleton equation. These are nearly identical to their word counterparts, but the chemical name is replaced by the chemical symbol. These equations also up their usefulness by

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state of matter
Equations
the reactants
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and products are
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