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What causes a chemical reaction

when you prepare scrambled eggs?

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Answer the following questions. (i)

What type of chemical reaction takes place in each of the two given steps?

(ii) Name the metal initially taken in the powder form. Write balanced chemical equations for both reactions. Answer.

(i) In first step, oxidation takes place. In second step, redox reaction takes

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Answer Type I. Question 1. Write the  
balanced chemical equation for the  
following reaction and identify the type  
of reaction and define it. 'Iron III oxide

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reacts with Aluminium and gives molten iron and aluminium oxide'.

(2012) Answer:  $\text{Fe}_2\text{O}_3 (\text{s}) + 2\text{Al} (\text{s}) \rightarrow 2\text{Fe} (\text{l}) + \text{Al}_2\text{O}_3 (\text{l}) + \text{Heat}$

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from Chapter 1 - Chemical Reactions and Equations of CBSE Class 10 Science. All these questions are based on important topics and concepts from the chapter.

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MCQs from Class 10 Science ...*

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Chemical reactions have the same number of atoms before the reaction as after the reaction. Balancing chemical equations is a basic skill in chemistry and testing yourself helps retain important information. This collection of ten chemistry test questions will give you practice in how

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These important questions help students in clearing their doubts so that they can score well in the exam.

*Important Questions For Class 10  
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Question: O CHEMICAL REACTIONS  
Writing Net Ionic Equations The

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Following Chemical Reaction Takes Place In Aqueous Solution:

$\text{CuCl}_2(\text{aq}) + 2 \text{NaOH}(\text{aq}) \rightarrow \text{Cu}(\text{OH})_2(\text{s}) + 2 \text{NaCl}(\text{aq})$  Write The Net Ionic Equation For This Reaction. 0-0 ? \$ ? This problem has been solved! See the answer.

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Questions Answers  
give any information about the speed of the reaction. Some chemical equations and reactions have diverse affect. Students likely find difficulty in balancing chemical equations worksheet. To help you resolve this issue, we have balancing equations worksheet with answers on our main

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chemical formula of a compound  
indicates its constituting elements and

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the number of atoms of each combining element. In molecular compounds, the combining capacity, i.e., valency of each combining element determines the chemical formula of the compound.

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Chemical Reactions and Equations

Extra Questions Long Answer Type.

Question 1. Consider the chemical equation given below and answer the questions that follow: (i) Name the substance which is getting oxidised.

(ii) Name the substance which is



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getting reduced. (iii) Name the oxidising agent. (iv) Name the reducing agent.

*Chemical Reactions and Equations  
Class 10 Extra Questions ...*

Preview this quiz on Quizizz. During a lab experiment, a scientist numbers

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three beakers. Next, she pours 150 mL of liquid into each beaker. Each liquid is different. Then, the scientist adds 5 g of white solid baking soda ( $\text{NaHCO}_3$ ) to each beaker and stirs. She observes the results. In which of the beakers does the scientist observe clear evidence of a chemical

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Quiz - Quizizz*

Chemical reactions and equations  
MCQ/objective questions Class 10  
Science Chapter 1 1. Magnesium  
ribbon is cleaned before burning.

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Why? A. It absorbs moisture from the atmosphere which interferes with burning. B. It's outer layer form oxides which interfere with burning. C. Dust from the atmosphere prevents its burning. D. All of the above

*Chemical reactions and equations*

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## MCQ - Study Answers

Martin Leigh / Getty Images Good job!

You completed the quiz, so you've seen examples of the different types of chemical reactions. If you're still a bit shaky on how to tell the types of reactions apart or if you just want more examples, you can review the main

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Reaction types. If you're ready to try another quiz, see how familiar you are with the units of measurement.

*Chemical Reaction Classification Quiz*  
Online Test of Chapter 1 Chemical Reactions and Equations 1 Science| Class 10th. Q1. A chemical reaction

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has taken place in which of the following process? (i) Ice melts into water (ii) A wet shirt got dried in sunlight (iii) A brown layer is formed over iron rod kept in air (iv) Sugar getting dissolved in water. Q2.

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Balancing chemical equations.

Balancing more complex chemical  
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balancing chemical equations.

Balancing another combustion reaction. Balancing chemical equation with substitution. Practice: Balancing chemical equations 1.

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